Solutions to JEE(Main)-2019

9th January 2019 (Second Shift)

PART -A (PHYSICS)

1.			•
	The ratio $\frac{R_B}{R_A}$ of their activities. The	e ratio $\frac{R_B}{R_A}$ of their activates after time	e t itself decays
	with time t as e ^{-3t} . If the half-life of A	is ℓ n2, the half-life of B is:	

At a given instant, say t = 0, two radioactive substances A and B have equal activates.



2. A power transmission line feeds input power at 2300 V to a step down transformer with its primary windings having 4000 turns. The output power is delivered at 230 V by the transformer. If the current in the primary of the transformer is 5A and its efficiency is 90%, the output current would be:

(A) 50 A (B) 45 A (C) 35 A (D) 25 A

3. The energy associated with electric field is (U_E) and with magnetic field is (U_B) for an electromagnetic wave in free space. Then:

(A) $U_{E} = \frac{U_{B}}{2}$ (B) $U_{E} > U_{B}$ (C) $U_{E} < U_{B}$ (D) $U_{E} = U_{B}$

4. A force acts on a 2 kg object so that its position is given as a function of time as $x = 3t^2 + 5$. What is the work done by this force in first 5 seconds?

(A) 850 J (B) 950 J (C) 875 J (D) 900 J

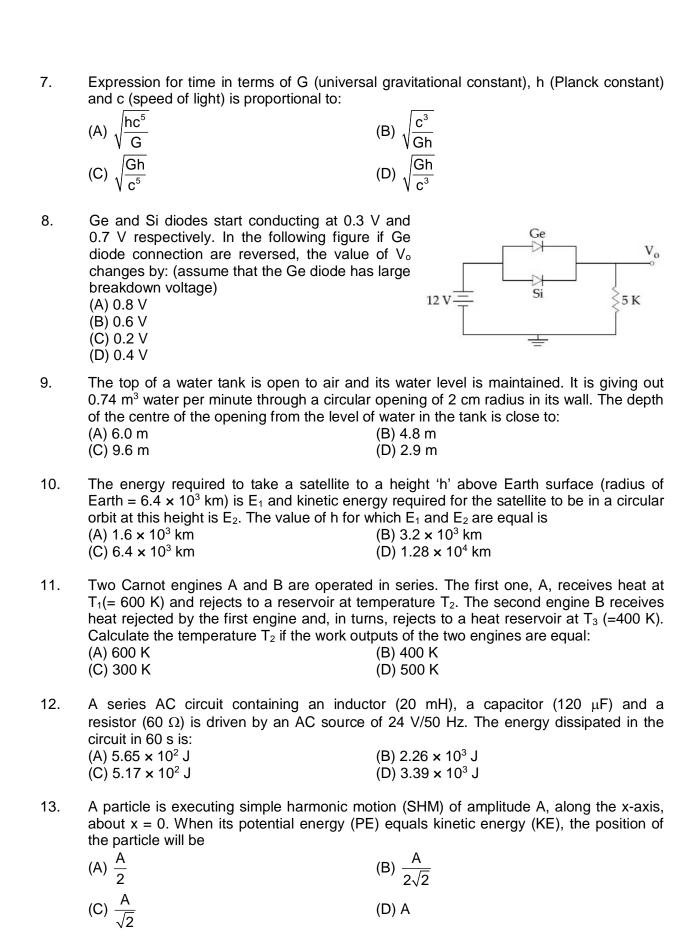
5. A particle having the same charge as of electron moves in a circular path of radius 0.5 cm under the influence of a magnetic field of 0.5 T. If an electric field of 100 V/m makes it to move in a straight path, then the mass of the particle is (given charge of electron = 1.6×10^{-19} C)

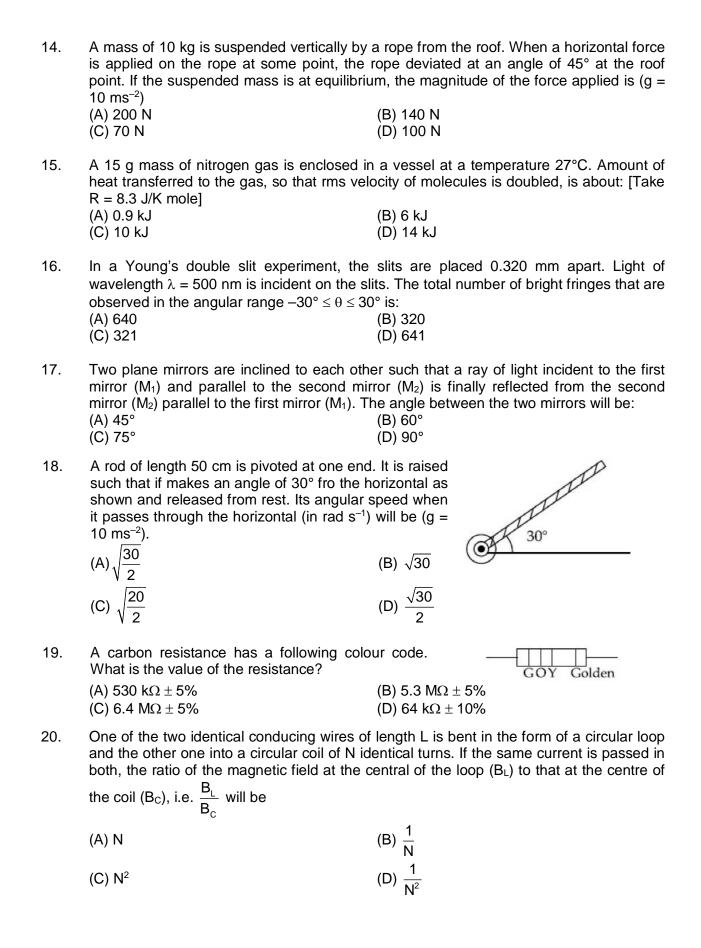
(A) 9.1×10^{-31} kg (B) 1.6×10^{-27} kg (C) 1.6×10^{-19} kg (D) 2.0×10^{-24} kg

6. Two point charges $q_1(\sqrt{10} \mu C)$ and $q_2(-25 \mu C)$ are placed on the x-axis at x=1 m and x=4 m respectively. The electric field (in V/m) at a point y=3 m on y-axis is,

$$\left[take \frac{1}{4\pi\epsilon_0} = 9 \times 10^9 Nm^2 C^{-2} \right]$$

(A) $(63\hat{i} - 27\hat{i}) \times 10^2$ (B) $(-63\hat{i} + 27\hat{i}) \times 10^2$ (C) $(81\hat{i} - 81\hat{i}) \times 10^2$ (D) $(-81\hat{i} + 81\hat{i}) \times 10^2$





- 21. A rod of mass 'M' and length '2L' is suspended at its middle by a wire. It exhibits torsional oscillations; If two masses each of 'm' are attached at distance 'L/2' from its centre on both sides, it reduces the oscillation frequency by 20%. The value of ratio m/M is close to:
 - (A) 0.77

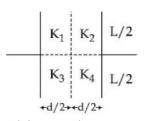
(C) 0.37

- (B) 0.57 (D) 0.17
- Charge is distributed within a sphere of radius R with a volume charge density 22. $\rho(r) = \frac{A}{r^2} e^{-2r/a}$, where A and a are constants. If Q is the total charge of this charge distribution, the radius R is:
 - (A) a $\log \left(1 \frac{Q}{2\pi a \Delta}\right)$

(B) $\frac{a}{2} \log \left| \frac{1}{1 - \frac{Q}{2 - \alpha A}} \right|$

(C) $\frac{a}{2} \log \left(\frac{1}{1 - \frac{Q}{Q}} \right)$

- (D) $\frac{a}{2} \log \left(1 \frac{1}{2\pi aA} \right)$
- 23. A parallel palate capacitor with square plates is filled with four dielectrics of dielectric constants K_1 , K_2 , K_3 , K_4 arranged as shown in the figure. The effective dielectric constant K will be:



(A) $K = \frac{(K_1 + K_3)(K_2 + K_4)}{K_1 + K_2 + K_3 + K_4}$ (C) $K = \frac{(K_1 + K_2)(K_3 + K_4)}{K_1 + K_2 + K_3 + K_4}$

(B) $K = \frac{(K_1 + K_2)(K_3 + K_4)}{2(K_1 + K_2 + K_3 + K_4)}$ (D) $K = \frac{(K_1 + K_4)(K_2 + K_3)}{2(K_1 + K_2 + K_3 + K_4)}$

- 24. The pitch and the number of divisions, on the circular scale, for a given screw gauge are 0.5 mm and 100 respectively. When the screw gauge is fully tightened without any object, the zero of its circular scale lies 3 divisions below the mean line.

The readings of the main scale and the circular scale for a thin sheet, are 5.5 mm and 48 respectively, the thickness of this sheet is

(A) 5.755 mm

(B) 5.950 mm

(C) 5.725 mm

- (D) 5.740 mm
- 25. A musician using an open flute of length 50 cm produces second harmonic sound

A person runs towards the musician from another end of a hall at a speed of 10 km/h. If the wave speed is 330 m/s, the frequency heard by the running person shall be close to:

(A) 666 Hz

(B) 753 Hz

(C) 500 Hz

(D) 333 Hz

26. In a car race on straight road, car A takes a time 't' less than car B at the finish and passes finishing point with a speed 'v' more than that of car B. Both the cars start from rest and travel with constant acceleration a₁ and a₂ respectively. Then 'v' is equal to

(A)
$$\frac{2a_1a_1}{a_1+a_2}t$$

(B)
$$\sqrt{2a_1a_2}$$
 t

(C)
$$\sqrt{a_1 a_2}$$
 t

(D)
$$\frac{a_1 + a_2}{2}t$$

27. The magnetic field associated with a light wave is given, at the origin, by

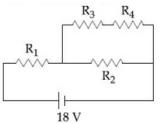
If this light falls on a silver plate having a work function of 4.7 eV, what will be the maximum kinetic energy of the photo electrons?

28. In the given circuit the internal resistance of the 18 V cell is negligible. If $R_1=400~\Omega,~R_3=100~\Omega$ and $R_4=500~\Omega$ and the reading of an ideal voltmeter across R_4 is 5 V, then the value of R_2 will be



(C) 550
$$\Omega$$





29. In a communication system operating at wavelength 800 nm, only one percent of source frequency is available as signal bandwidth. The number of channels accommodated for transmitting TV signals of band width 6 MHz are (Take velocity of light $c = 3 \times 10^8$ m/s, $h = 6.6 \times 10^{-34}$ J-s)

(A)
$$3.75 \times 10^6$$

(B)
$$3.86 \times 10^6$$

(C)
$$6.25 \times 10^5$$

$$(D)$$
 4.87 × 10⁵

30. The position co-ordinates of a particle moving in a 3-D coordinates system is given by

$$x = a \cos \omega t$$

$$y = a \sin \omega t$$

and
$$z = a\omega t$$

The speed of the particle is:

(A)
$$\sqrt{2}$$
 a ω

(C)
$$\sqrt{3}$$
 a ω

PART -B (CHEMISTRY)

- The entropy change associated with the conversion of 1 kg of ice at 273 K to water 31. vapours at 383 K is: (Specific heat of water liquid and water vapours are 4.2 kJ K⁻¹ kg⁻¹ and 2.0 kJ K⁻¹ kg⁻¹, heat of liquid fusion and vapourisation of water are 334 kJ kg⁻¹ and 2491 kJ kg⁻¹, respectively) (log 273 = 2.436, log 373 = 2.572, log 383 = 2.583)
 - (A) 7.90 kJ K⁻¹ kg⁻¹

(B) 2.64 kJ K⁻¹ kg⁻¹ (D) 9.26 kJ K⁻¹ kg⁻¹

(C) 8.49 kJ K⁻¹ kg⁻¹

- For the following reaction the mass of water produced from 445 g of C₅₇H₁₁₀O₆ is 32. $2C_{57}H_{110}O_{6}(s) + 163O_{2}(g) \longrightarrow 114CO_{2}(g) + 110H_{2}O(l)$
 - (A) 490 g

(B) 445 g

(C) 495 g

- (D) 890 g
- The major product formed in the following reaction is: 33.

(A)

- (B)
- (C)
- (D)
- 34. Which of the following conditions in drinking water causes methemoglobinemia?
 - (A) > 50 ppm of lead

(B) > 50 ppm of chloride

(C) > 50 ppm of nitrate

- (D) > 100 ppm of sulphate
- 35. The major product of the following reaction is:

OH CH₃ AlCl₃,
$$\Delta$$

 CH_3

(A)

(B)

(C)

OH CH_3 (D)

36. The major product obtained in the following reaction is:

OH
$$(CH_3CO)_2O$$
/pyridine (1 eqv.) room temp. NH_2

- (A) NHCOCH₃
- COCH₃ ŎН (B) $\dot{N}H_2$

(D)

OCOCH₃ (C)

- OH NHCOCH₃
- 37. The major product of the following reaction is:

$$\begin{array}{c|c}
O \\
C \\
NH_2 \\
CH_2CH_3
\end{array}$$
(i) Br₂/hv
(ii) KOH (dil)

NH (A)

(B) NH

NH (C) CH_3

- NH (D)
- 38. The correct match between item I and item II is

Item II

- Benzaldehyde (a)
- Alumina (b)
- Acetonitrile (c)
- (A) $a \rightarrow q$, $b \rightarrow p$, $c \rightarrow r$
- (C) $a \rightarrow q$, $b \rightarrow r$, $c \rightarrow p$
- Mobile phase (p) Adsorbent (q)
- (r) Adsorbate
 - - (B) $a \rightarrow r$, $b \rightarrow q$, $c \rightarrow p$
 - (D) $a \rightarrow p, b \rightarrow r, c \rightarrow q$
- 39. The metal that forms nitride by reacting directly with N₂ of air is
 - (A) K

(B) Li

(C) Rb

- (D) Cs
- 40. For coagulation of arsenious sulphide sol, which one of the following salt solution will be most effective?
 - (A) BaCl₂

(B) AICI₃

(C) NaCl

(D) Na₃PO₄

- 41. The complex that has highest crystal field splitting energy(Δ) is
 - (A) $[Co(NH_3)_5(H_2O)]CI_3$

(B) $K_2[CoCl_4]$

(C) [Co(NH₃)₅Cl]Cl₂

- (D) $K_3[Co(CN)_6]$
- 42. The pH of rain water is approximately
 - (A) 5.6

(B) 7.5

(C) 7.0

- (D) 6.5
- 43. Consider the following reversible chemical reactions:

$$A_2(g) + B_2(g) \xrightarrow{\kappa_1} 2AB(g)$$
(1)

$$6 AB(g) \stackrel{\kappa_2}{\rightleftharpoons} 3 A_2(g) + 3 B_2(g) \dots (2)$$

The relation between K₁ and K₂ is

(A)
$$K_1K_2 = \frac{1}{3}$$

(B)
$$K_2 = K_1^3$$

(C)
$$K_2 = K_1^{-3}$$

(D)
$$K_1K_2 = 3$$

44. The correct sequence of amino acids present in the tripeptide given below is

$$\begin{array}{c|c} Me & Me & OH \\ H_2N & N & OH \\ OH & OH \\ OH & OH \\ \end{array}$$

(A) Val – Ser – Thr

(B) Thr - Ser - Val

(C) Leu - Ser - Thr

- (D) Thr Ser Leu
- 45. For the reaction, $2A + B \longrightarrow products$, when the concentrations of A and B both were doubled, the rate of the reaction increased from 0.3 mol L^{-1} s⁻¹ to 2.4 mol L^{-1} s⁻¹. When the concentration of A alone is doubled, the rate increased from 0.3 mol L^{-1} s⁻¹ to 0.6 mol L^{-1} s⁻¹.

Which one of the following statements is correct?

- (A) Total order of the reaction is 4
- (B) Order of the reaction with respect to B is 2
- (C) Order of the reaction with respect to B is 1
- (D) Order of the reaction with respect to A is 2
- 46. The products formed in the reaction of cumene with O_2 followed by treatment with dil. HCl are:

(A) and
$$H_3C$$
 CH_3

47. The tests performed on compound X and their inferences are:

Test Interference (a) 2, 4-DNP test Colorued precipitate (b) Iodoform test Yellow precipitate (c) Azo-dye test No dye formation

Compound 'X' is $H_3C CH_3$ (A) CHO $NH_2 OH$ (C) CH_3 (D) CHO

48. If the standard electrode potential for a cell is 2 V at 300 K, the equilibrium constant (K) for the reaction

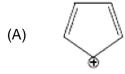
$$Zn(s) + Cu^{2+}(aq) \Longrightarrow Zn^{2+}(aq) + Cu(s)$$

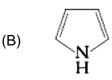
At 300 K is approximately

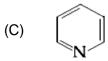
 $(R = 8 \text{ JK}^{-1} \text{ mol}^{-1}, F = 96000 \text{ C mol}^{-1})$

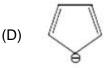
- (A) e^{-80} (B) e^{-160} (C) e^{320} (D) e^{160}
- 49. The temporary hardness of water is due to
 - $\begin{array}{lll} \text{(A) Na}_2\text{SO}_4 & \text{(B) NaCl} \\ \text{(C) Ca}(\text{HCO}_3)_2 & \text{(D) CaCl}_2 \\ \end{array}$
- 50. In which of the following processes, the bond order has increased and paramagnetic character has changed to diamagnetic?
 - (A) NO \longrightarrow NO⁺ (B) N₂ \longrightarrow N₂⁺ (C) O₂ \longrightarrow O₂⁻ (D) O₂ \longrightarrow O₂²⁻
- 51. Which of the following combination of statements is true regarding the interpretation of the atomic orbitals?
 - (1) An electron in an orbital of high angular momentum stays away from the nucleus than an electron in the orbital of lower angular momentum.
 - (2) For a given value of the principal quantum number, the size of the orbit is inversely proportional to the azimuthal quantum number
 - (3) According to wave mechanics, the ground state angular momentum is equal to $\frac{h}{2\pi}$
 - (4) The plot of ψ Vs r for various azimuthal quantum numbers, shows peak shifting towards higher r value
 - (A) (1), (4) (C) (1), (3) (B) (1), (2) (D) (2), (3)

52. Which of the following compounds is not aromatic?







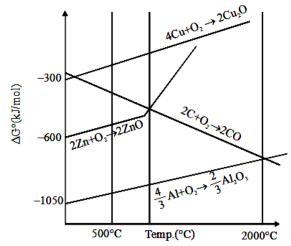


- 53. Good reducing nature of H₃PO₂ is attributed to the presence of
 - (A) Two P OH bonds

(B) One P - H bond

(C) Two P – H bonds

- (D) One P OH bond
- 54. The correct statement regarding the given Ellingham diagram is



- (A) At 1400°C, Al can be used for the extraction of Zn from ZnO
- (B) At 500°C, coke can be used for the extraction of Zn from ZnO
- (C) Coke cannot be used for the extraction of Cu from Cu₂O
- (D) At 800°C Cu can be used for the extraction of Zn from ZnO
- 55. The transition element that has lowest enthalpy of atomisation is
 - (A) Fe

(B) Cu

(C) V

- (D) Zn
- 56. The increasing basicity order of the following compounds is
 - (1) $CH_3CH_2NH_2$

(2) CH_2CH_3 CH_3CH_2NH

 $\begin{array}{ccc}
& & \text{CH}_3 \\
\text{H}_3\text{C}-\text{N}-\text{CH}_2
\end{array}$

 $\begin{array}{ccc}
 & CH_3 \\
 & | \\
 Ph-N-H
\end{array}$

(A) (4) < (3) < (2) < (1)

(B) (4) < (3) < (1) < (2)

(C) (1) < (2) < (3) < (4)

(D) (1) < (2) < (4) < (3)

57.	When the first electron gain enthalpy ($\Delta_{eg}H$) of oxygen is -141 kJ/mol, its second electron gain enthalpy is					
	(A) a more negative value than the first(C) negative, but less negative than the first	(B) almost the same as that of the first st (D) a positive value				
58.	At 100°C, copper(Cu) has FCC unit cell structure with cell edge length $x \stackrel{\circ}{A}$. What is the approximate density of Cu(in g cm ⁻³) at this temperature? [Atomic mass of Cu = 63.55 u]					
	(A) 205	_(D) 105				



59. A solution containing 62 g ethylene glycol in 250 g water is cooled to -10 $^{\circ}$ C. If K_f for water is 1.86 K kg mol⁻¹, the amount of water(in g) separated as ice is (A) 48 (B) 32 (C) 64 (D) 16

60. Homoleptic octahedral complexes of a metal ion 'M³+' with three monodentate ligands L₁, L₂ and L₃ absorb wavelengths in the region of green, blue and red respectively. The

increasing order of the ligand strength is (A) $L_3 < L_1 < L_2$ (B) $L_3 < L_2 < L_1$ (C) $L_1 < L_2 < L_3$ (D) $L_2 < L_1 < L_3$

PART-C (MATHEMATICS)

61. The sum of the following series

 $1+6+\frac{9 \left(1^2+2^2+3^2\right)}{7}+\frac{12 \left(1^2+2^2+3^2+4^2\right)}{9}+\frac{15 \left(1^2+2^2+....+5^2\right)}{11}+... \text{ up to 15 terms, is:}$

(A) 7820

(C) 7520

- (B) 7830
- (D) 7510
- 62. For each $x \in R$, let [x] be the greatest integer less than or equal to x. Then $\lim_{x \to 0^+} \frac{x\left([x] + |x|\right)\sin[x]}{|x|} \text{ is equal to}$
 - $(A) \sin 1$

(B) 0

(C) 1

- (D) sin 1
- 63. Let $f:[0,1] \to R$ be such that f(xy) = f(x) f(y) for all $x,y \in [0,1]$, and $f(0) \neq 0$. If y = y(x) satisfies the differential equation, $\frac{dy}{dx} = f(x)$ with y(0) = 1, then $y(\frac{1}{4}) + y(\frac{3}{4})$ is equal to
 - (A) 4

(B) 3

(C) 5

- (D) 2
- 64. If $x = \sin^{-1}(\sin 10)$ and $y = \cos^{-1}(\cos 10)$, then y x is equal to:
 - (A) π

(B) 7π

(C) 0

- (D) 10
- 65. If $0 \le x < \frac{\pi}{2}$, then the number of values of x for which $\sin x \sin 2x + \sin 3x = 0$, is
 - (A) 2

(B)

(C) 3

- (D) 4
- 66. Let z_0 be a root of the quadratic equation, $x^2 + x + 1 = 0$. If $z = 3 + 6iz_0^{81} 3iz_0^{93}$, then arg z is equal to
 - (A) $\frac{\pi}{4}$

(B) $\frac{\pi}{3}$

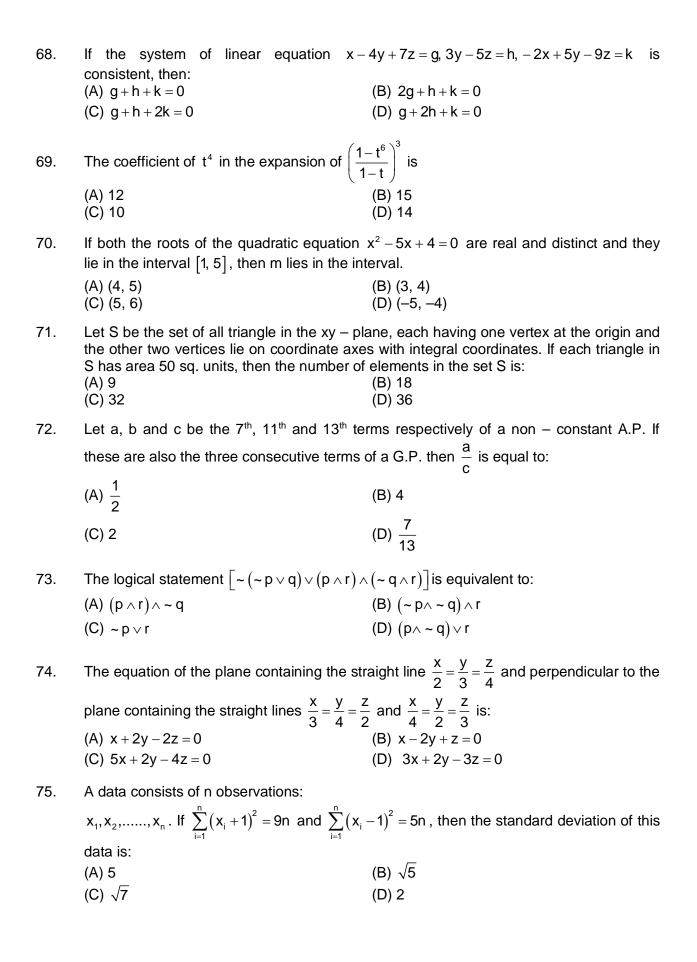
(C) 0

- (D) $\frac{\pi}{6}$
- 67. The area of the region $A\{(x,y):0 \le y \le x |x| + 1 \text{ and } -1 \le x \le 1\}$ in sq. units, is:
 - (A) $\frac{2}{3}$

(B) $\frac{1}{3}$

(C) 2

(D) $\frac{4}{3}$



76. If
$$A = \begin{bmatrix} e^t & e^{-t} \cos t & e^{-t} \sin t \\ e^t & -e^{-t} \cos t - e^{-t} \sin t & -e^{-t} \sin t + e^{-t} \cos t \\ e^t & 2e^{-t} \sin t & -2e^{-t} \cos t \end{bmatrix}$$
 Then A is

(A) Invertible only if $t = \frac{\pi}{2}$

(B) not invertible for any $t \in R$

(C) invertible for all $t \in R$

(D) invertible only if $t = \pi$

77. If
$$f(x) = \int \frac{5x^8 + 7x^6}{(x^2 + 1 + 2x^7)^2} dx$$
, $(x \ge 0)$ and $f(0) = 0$, then the value of $f(1)$ is:

(A) $-\frac{1}{2}$

(B) $\frac{1}{2}$

(C) $-\frac{1}{4}$

(D) $\frac{1}{4}$

78. Let f be a differentiable function R to R such that
$$\left|f\left(x\right)-f\left(y\right)\right|\leq 2\left|x-y\right|^{\frac{3}{2}}$$
, for all $x,y\in R$. If $f\left(0\right)=1$ then $\int\limits_{0}^{1}f^{2}\left(x\right)dx$ is equal to

(A) 0

(B) $\frac{1}{2}$

(C) 2

(D) 1

79. If
$$x = 3 \tan t$$
 and $y = 3 \sec t$, then the value of $\frac{d^2y}{dx^2}$ at $t = \frac{\pi}{4}$, is:

 $(A) \ \frac{3}{2\sqrt{2}}$

(B) $\frac{1}{3\sqrt{2}}$

(C) $\frac{1}{6}$

(D) $\frac{1}{6\sqrt{2}}$

80. The number of natural numbers less than 7,000 which can be formed by using the digits 0, 1, 3, 7, 9 (repetition of digits allowed) is equal to:

(A) 250

(B) 374

(C) 372

(D) 375

81. If the circles $x^2 + y^2 - 16x - 20y + 164 = r^2$ and $(x-4)^2 + (y-7)^2 = 36$ intersect at two distinct points, then:

(A) 0 < r < 1

(B) 1 < r < 11

(C) r > 11

(D) r = 11

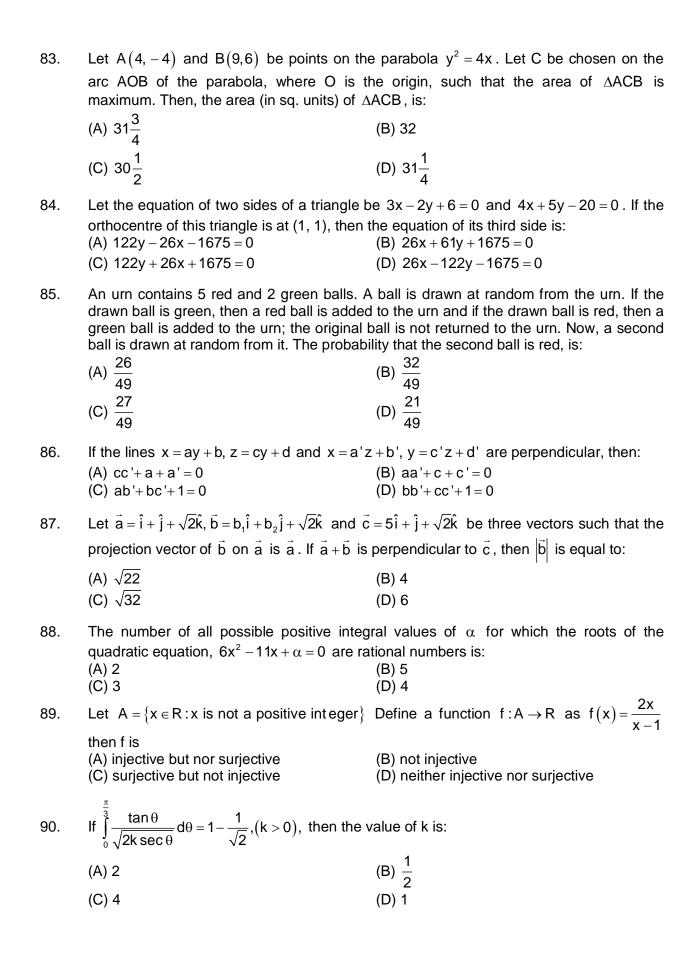
82. A hyperbola has its centre at the origin, passes through the point (4, 2) and has transverse axis of length 4 along the x – axis. Then the eccentricity of the hyperbola is:

(A) $\frac{2}{\sqrt{3}}$

(B) $\frac{3}{2}$

(C) √3

(D) 2



JEE (Main) – 2019 ANSWERS

PART A - PHYSICS

1A(1A - 1113103											
С	2.	В	3.	D	4.	D					
D	6.	Α	7.	С	8.	D					
В	10.	В	11.	D	12.	С					
С	14.	D	15.	С	16.	D					
В	18.	В	19.	Α	20.	D					
С	22.	В	23.	Α	24.	С					
Α	26.	С	27.	D	28.	Α					
С	30.	Α									
PART B – CHEMISTRY											
D	32.	С	33.	С	34.	С					
С	36.	D	37.	С	38.	В					
В	40.	В	41.	D	42.	Α					
С	44.	Α	45.	В	46.	С					
В	48.	D	49.	С	50.	Α					
Α	52.	Α	53.	С	54.	Α					
D	56.	В	57.	D	58.	D					
С	60.	Α									
				. =:00							
	PAF	RT C – MAT	HEM	ATICS							
Α	62.	Α	63.	В	64.	Α					
Α	66.	Α	67.	С	68.	В					
В	70.	(Bonus)	71.	D	72.	В					
Α	74.	В	75.	В	76.	С					
D	78.	D	79.	D	80.	В					
В	82.	Α	83.	D	84.	D					
В	86.	В	87.	D	88.	С					
Α	90.	Α									
	D B C B C A C D C B A D C A A B A D B B	C 2. D 6. B 10. C 14. B 18. C 22. A 26. C 30. PART A 62. A 66. B 70. A 74. D 78. B 82. B 86.	C 2. B D 6. A B 10. B C 14. D B 18. B C 22. B A 26. C C 30. A PART B - CH D 32. C C 36. D B 40. B C 44. A B 48. D A 52. A D 56. B C 60. A PART C - MAT A 62. A A 66. A B 70. (Bonus) A 74. B D 78. D B 82. A B 86. B	C 2. B 3. C 7. B 11. C 14. D 15. B 19. C 22. B 23. A 26. C 27. C 30. A PART B - CHEMIS D 37. B 40. B 41. C 44. A 45. B 48. D 49. A 52. A 53. D 56. B 57. C 60. A PART C - MATHEMA A 62. A 63. A 66. A 67. B 70. (Bonus) 71. A 74. B 75. D 78. D 79. B 82. A 83. B 86. B 87.	C 2. B 3. D D 6. A 7. C B 10. B 11. D C 14. D 15. C B 18. B 19. A C 22. B 23. A A 26. C 27. D C 30. A The match stry D 32. C 33. C C 36. D 37. C B 40. B 41. D C 44. A 45. B B 48. D 49. C A 52. A 53. C D 56. B 57. D C 60. A 67. C B 70. (Bonus) 71. D A 66. A 67. C B 70. (Bonus) 71.	C 2. B 3. D 4. D 6. A 7. C 8. B 10. B 11. D 12. C 14. D 15. C 16. B 18. B 19. A 20. C 22. B 23. A 24. A 26. C 27. D 28. C 30. A A 24. PART B – CHEMISTRY D 32. C 33. C 34. C 36. D 37. C 38. B 40. B 41. D 42. C 44. A 45. B 46. B 48. D 49. C 50. A 52. A 53. C 54. D 56. B 57. D 58. C 60. A 67. <					

HINTS AND SOLUTIONS

PART A-PHYSICS

1.
$$R = R_{o} e^{-\lambda t}$$

$$\therefore \frac{R_{B}}{R_{A}} = \frac{R_{o} e^{-\lambda_{B} t}}{R_{o} e^{-\lambda_{B} t}} = e^{-(\lambda_{B} - \lambda_{A})t} = e^{-3t}$$

$$\Rightarrow \lambda_{B} - \lambda_{A} = 3$$

$$\Rightarrow \frac{\ell n^{2}}{T_{B}} - \frac{\ell n2}{\ell n2} = 3.$$

$$\Rightarrow T_{B} = \frac{\ell n2}{4}$$

$$\begin{split} 2. \qquad & P_s = \eta \, P_P \\ \qquad & \Rightarrow \quad E_s \, \, i_s = \eta E_i i_P \\ \qquad & \Rightarrow \quad i_s = \frac{(0.9)\,(2300)\,(5)}{(230)} = 45 \,\, A. \end{split}$$

3.
$$B = \frac{E}{C}$$

$$\Rightarrow U_{E} = \frac{1}{2} \varepsilon_{o} E^{2}$$

$$U_{B} = \frac{B^{2}}{2\mu_{o}} = \frac{E^{2}}{2\mu_{o} C^{2}} = \frac{E^{2}}{2\mu_{o}} (\mu_{o} \varepsilon_{o}) = U_{E}$$

4.
$$x = 3t^2 + 5$$

 $\Rightarrow v = 6t$
 $\Rightarrow \Delta W = \Delta k$
 $= \frac{1}{2}(2)(30)^2 - \frac{1}{2}2(0)^2$
 $= 900 \text{ J}$

5.
$$eE = evB$$

$$\Rightarrow E = \left(\frac{eBr}{m}\right)B$$

$$\Rightarrow m = \frac{eB^2r}{E}$$

$$\Rightarrow m = \frac{(1.6 \times 10^{-19})(0.5)^2(0.5 \times 10^{-2})}{100} = 2 \times 10^{-24} \text{ kg.}$$

6.
$$\vec{E} = \frac{kq_1}{r_1^3} \vec{r_1} + \frac{kq_2}{r_2^3} \vec{r_2} = k \times 10^{-6} \left[\frac{\sqrt{10}}{10\sqrt{10}} (-\hat{i} + 3\hat{j}) + \frac{(-25)}{125} (-4\hat{i} + 3\hat{j}) \right]$$
$$= (9 \times 10^3) \left[\frac{1}{10} (-\hat{i} + 3\hat{j}) - \frac{1}{5} (-4\hat{i} + 3\hat{j}) \right]$$

$$= (9 \times 10^{3}) \left[\left(-\frac{1}{10} + \frac{4}{5} \right) \hat{\mathbf{i}} + \left(\frac{3}{10} - \frac{3}{5} \right) \hat{\mathbf{i}} \right] = 9000 \left(\frac{7}{10} \hat{\mathbf{i}} - \frac{3}{10} \hat{\mathbf{j}} \right)$$
$$= (63\hat{\mathbf{i}} - 27\hat{\mathbf{j}}) (100)$$

7.
$$\begin{aligned} t &= G^a \, h^b \, c^c \\ &\Rightarrow & M^o \, L^o \, T' = (M^{-1} \, L^3 \, T^{-2})^a \, (ML^2 T^{-1})^b \, (LT^{-1})^c \\ &\Rightarrow & -a + b = 0 \, \Rightarrow \, a = b \\ &\Rightarrow & 3a + 2b + c = 0 \\ &\Rightarrow & c = -5a \\ &\Rightarrow & -2a - b - c = 1 \\ &\Rightarrow & a = \frac{1}{2} \; ; \; b = \frac{1}{2} \; ; \; c = -\frac{5}{2} \end{aligned}$$

8.
$$V_{O_i} = 12 - 0.3 = 11.7 \text{ V}$$

 $V_{O_f} = 12 - 0.7 = 11.3 \text{ V}$
 $\Rightarrow \Delta V_{O} = -0.4 \text{ V}$

9.
$$\frac{dV}{dt} = Av \implies \frac{dV}{dt} = A\sqrt{2gh}$$

$$\Rightarrow \frac{0.74}{60} = (3.14) \left(\frac{2}{100}\right)^2 \sqrt{2(9.8)h}$$

$$\Rightarrow h = 4.92 \text{ m}$$

10.
$$E_1 = -\frac{GMm}{R+h} - \left(-\frac{GMm}{R}\right)$$

$$E_2 = \frac{1}{2}m\left(\sqrt{\frac{GM}{R+h}}\right)^2 = \frac{GMm}{2(R+h)}$$

$$E_1 = E_2 \quad ; \quad h = \frac{R}{2}$$

11.
$$W_1 = W_2$$

 $\Rightarrow 600 - T_2 = T_2 - 400$
 $\Rightarrow T_2 = 500 \text{ K}$

12.
$$E = Pt = \frac{E^2}{Z^2}Rt = \frac{(24)^2}{60^2 + (8.33\pi - 2\pi)^2}(60)(60) = 518 \text{ J}.$$

13. PE = KE

$$\Rightarrow \frac{1}{2}m\omega^{2}(A^{2} - x^{2}) = \frac{1}{2}m\omega^{2}x^{2}$$

$$\Rightarrow x = \frac{A}{\sqrt{2}}$$

14. T cos
$$45^{\circ}$$
 = mg
T sin 45° = F
 \Rightarrow F = mg = 100 N.

15.
$$\Delta Q = \frac{f}{2} nR\Delta T$$
$$= \frac{5}{2} \left(\frac{15}{28}\right) (8.3) (1200 - 300) = 10000 \text{ J}.$$

16.
$$\Delta X_{max} = d \sin \theta = 0.32 \sin 30 = 0.16 \text{ mm}$$

$$\therefore \quad n = \frac{\Delta X_{\text{max}}}{\lambda} = \frac{0.16 \times 10^{-3}}{500 \times 10^{-9}}$$
$$= \frac{0.16 \times 10^{6}}{500} = \frac{1600}{5} = 320$$

$$\therefore$$
 Number of BFs = $(2n + 1) = 641$

$$\theta = 60^{\circ}$$

18.
$$mg \frac{\ell}{2} \left(\frac{1}{2} \right) = \frac{1}{2} \left(\frac{m\ell^2}{3} \right) \omega^2$$

$$\Rightarrow \omega = \sqrt{\frac{3g}{2\ell}} = \sqrt{30}$$

19.
$$R = 530 \text{ k}\Omega \pm 5\%$$

20.
$$B_{L} = \frac{\mu_{o}i}{2R}$$

$$B_{C} = \frac{\mu_{o}Ni}{2(R/N)}$$

$$\therefore \frac{B_{L}}{B_{C}} = \frac{1}{N^{2}}$$

21.
$$f = \frac{1}{2\pi} \sqrt{\frac{C}{\left(\frac{ML^{2}}{3}\right)}} & 8 & 0.8 \text{ f} = \frac{1}{2\pi} \sqrt{\frac{C}{\left(\frac{ML^{2}}{3} + \frac{mL^{2}}{2}\right)}}$$
$$\Rightarrow \frac{25}{16} = \frac{\frac{ML^{2}}{3} + \frac{mL^{2}}{2}}{\frac{ML^{2}}{3}}$$

$$\Rightarrow \frac{25}{16} = 1 + \frac{3 \text{ m}}{2 \text{ M}}$$

$$\Rightarrow \frac{9}{16} = \frac{3 \text{ m}}{2 \text{ M}}$$

$$\Rightarrow \frac{m}{M} = \frac{3}{8} = 0.37$$

22.
$$Q = \int \rho 4\pi r^2 dr = \int_0^R \left(\frac{A}{r^2} e^{\frac{-2r}{a}}\right) (4\pi r^2) dr$$
$$= 4\pi A \frac{a}{2} \left(1 - e^{\frac{-2R}{a}}\right)$$
$$\Rightarrow R = \frac{-a}{2} log \left(1 - \frac{Q}{2\pi Aa}\right)$$

23.
$$C_{1} = \frac{\varepsilon_{0}K_{1}\frac{L^{2}}{2}}{\frac{d}{2}} + \frac{\varepsilon_{0}K_{3}\frac{L^{2}}{2}}{\left(\frac{d}{2}\right)} = \frac{\varepsilon_{0}L^{2}}{d}(K_{1} + K_{3})$$

$$C_{2} = \frac{\varepsilon_{0}K_{2}\frac{L^{2}}{2}}{\frac{d}{2}} + \frac{\varepsilon_{0}K_{4}\frac{L^{2}}{2}}{\frac{d}{2}} = \frac{\varepsilon_{0}L^{2}}{d}(K_{2} + K_{4})$$

$$\therefore \frac{1}{c} = \frac{1}{c_{1}} + \frac{1}{c_{2}}$$

$$\Rightarrow \frac{d}{\varepsilon_{0}KL^{2}} = \frac{d}{\varepsilon_{0}L^{2}(K_{1} + K_{3})} + \frac{d}{\varepsilon_{0}L^{2}(K_{2} + K_{4})}$$

24. Zero error =
$$0 + 3 \times \frac{0.5 \text{ mm}}{100} = 0.015 \text{ mm}$$

MSR = $5.5 + 48 \times \frac{0.5}{100}$
= 5.74 mm .
 \therefore Thickness = $5.74 - 0.015 = 5.725 \text{ mm}$

25.
$$f = \frac{2}{2\ell} v_s = \frac{330}{0.5} = 660 \text{ Hz}$$

$$\therefore f' = f \left(\frac{v_s + v}{v_s} \right) = (660) \left(\frac{330 + \frac{50}{18}}{330} \right) = 660 \left(1 + \frac{50}{18 \times 330} \right)$$

$$= 666 \text{ Hz}.$$

$$26. \qquad \sqrt{\frac{2\ell}{a_2}} - \sqrt{\frac{2\ell}{a_1}} = t \qquad \qquad \Rightarrow \qquad \frac{\sqrt{2\ell}}{t} = \frac{\sqrt{a_1 a_2}}{\sqrt{a_1} - \sqrt{a_2}}$$

$$\sqrt{2a_1\ell} - \sqrt{2a_2\ell} = v \qquad \Rightarrow \qquad \frac{\sqrt{2\ell}}{v} = \frac{1}{\sqrt{a_1} - \sqrt{a_2}}$$

$$\Rightarrow \qquad \frac{v}{t} = \sqrt{a_1 a_2} \qquad \Rightarrow \qquad v = (\sqrt{a_1 a_2}) t$$

27.
$$KE_{max} = h\nu_{max} - \phi$$

$$= \frac{(6.6 \times 10^{-34}) (6.28 \times 10^{7}) (3 \times 10^{8})}{1.6 \times 10^{-19} \times 2 \times 3.14} - 4.7$$

$$= 12.37 - 4.7 = 7.67 \text{ eV}$$

28.
$$\frac{12}{400} = \frac{6}{600} + \frac{6}{R_{2}}$$

$$\Rightarrow \frac{1}{200} = \frac{1}{600} + \frac{1}{R_{2}}$$

$$\Rightarrow R_{2} = 300 \Omega$$

29.
$$f = \frac{c}{\lambda} = \frac{3 \times 10^8}{8 \times 10^{-7}} = \frac{3}{8} \times 10^{15} \text{Hz}$$

$$\therefore \qquad n = \frac{(0.01) \text{ f}}{6 \times 10^6} = \frac{\frac{3}{8} \times 10^{13}}{6 \times 10^6}$$

$$= \frac{1}{16} \times 10^7 = 6.25 \times 10^5$$

30.
$$v_{x} = \frac{dx}{dt} = -a\omega \sin \omega t$$

$$v_{y} = \frac{dy}{dt} = a\omega \cos \omega t$$

$$v_{z} = \frac{dz}{dt} = a\omega$$

$$\therefore v = \sqrt{v_{x}^{2} + v_{y}^{2} + v_{z}^{2}} = a\omega \sqrt{2}$$

PART B – CHEMISTRY

31.
$$H_2O(s) \longrightarrow H_2O(\ell) \longrightarrow H_2O(\ell) \longrightarrow H_2O(g) \longrightarrow H_2O(g)$$

1 kg 1 kg 1 kg
at 273 K at 273 K at 373 K at 373 K at 383 K
$$\Delta S = \Delta S_1 + \Delta S_2 + \Delta S_3 + \Delta S_4$$

$$= \frac{334}{273} + 4.2\ell n \frac{373}{273} + \frac{2491}{373} + 2\ell n \frac{383}{373} = 9.267 \text{ kJ Kg}^{-1} \text{ K}^{-1}$$

32.
$$2C_{57}H_{110}O_{6}(s) + 163O_{2}(g) \longrightarrow 114CO_{2}(g) + 110H_{2}O(I)$$

$$\frac{\text{Moles of } C_{57}H_{110}O_{6}}{2} = \frac{\text{Moles of } H_{2}O}{110}$$

$$\frac{\frac{445}{890}}{2} = \frac{\frac{\text{Mass of } H_{2}O}{18}}{110}$$

$$\text{Mass of } H_{2}O = 495 \text{ g}$$

33.
$$\begin{array}{c|c} O & O & O^- \\ || & \bigcirc & || \\ Ph-C-CH_3 & \xrightarrow{NaOH} & \Big[Ph-C-CH_2 \longleftrightarrow Ph-C=CH_2\Big] \\ \\ & \text{enolate ion} \end{array}$$

34. Fact based

35.

36. Nucleophilicity of NH₂> OH

$$\begin{array}{c|c} \text{OH} & \text{OH} \\ \hline \\ \text{NH}_2 & \xrightarrow{\text{(CH}_3\text{CO)}_2\text{O/Py}} & \text{O} \\ \hline \\ \text{NH}_2 & \text{NHCOCH}_3 \\ \end{array}$$

- 38. Acetonitrile is used as mobile phase for most of the reverse chromatography. Benzaldehyde is adsorbed on alumina.
- 39. The only alkali metal which forms nitride by reacting directly with N₂ is 'Li'.
- 40. As₂S₃ is a negatively charged sol. so AlCl₃ will be most effective.
- 41. As CN⁻ is a strong field ligand. K₃[Co(CN)₆] will have maximum 'Δ'.
- 42. Fact based.

43.
$$A_{2}(g) + B_{2}(g) \xrightarrow{K_{1}} 2 AB(g) \qquad \dots \dots (1)$$

$$6 AB(g) \xrightarrow{K_{2}} 3 A_{2}(g) + 3 B_{2}(g) \qquad \dots \dots (2)$$

$$Reaction(2) = -3 \times reaction(1)$$

$$\therefore K_{2} = \left(\frac{1}{K_{1}}\right)^{3} \Rightarrow K_{2} = K_{1}^{-3}$$

44.

$$\begin{array}{c|c} Me & Me & OH \\ H_2N & & & \\ OH & & \\ OH & & \\ \hline \\ OH & \\ \hline \\ H_2O & \\ \end{array}$$

45. $2A + B \longrightarrow products$

Rate =
$$K[A]^x[B]^y$$

$$r = K[A]^x[B]^y - - - - (i)$$

$$0.3 = K[A]^{x}[B]^{y} - - - (1)$$

$$2.4 = K[2A]^{x}[2B]^{y} - - - (2)$$

$$0.6 = K[2A]^{x}[B]^{y} - - - (3)$$

$$x = 1, y = 2$$

 H_3C

Overall order =
$$2 + 1 = 3$$

Order w.r.t
$$A = 1$$

Order w.r.t
$$B = 2$$

46.

47. ∵ -COCH₃ is present it will show both 2, 4-DNP & iodoform test.

Due to steric inhibition of resonance. I.P of 'N' is not involved in delocalization so coupling reaction will not take place.

48.
$$Zn(s) + Cu^{2+}(aq) \rightleftharpoons Zn^{2+}(aq) + Cu(s)$$

$$-nFE_{cell} = -RT\ell nK$$

$$\ell$$
nK = $\frac{2 \times 96500 \times 2}{9 \times 200}$ = 160.83

$$K = e^{160}$$

49. Fact based.

50.
$$NO \longrightarrow NO^+$$
 $N_2 \longrightarrow N_2^+$

B.O 0.5 3 B.O 3 2.5

Para Dia Dia Para

 $O_2 \longrightarrow O_2^+$ $O_2 \longrightarrow O_2^{2-}$

B.O 2 2.5 B.O 2 1

Para Para Dia

- 51. Refer Theory
- 52. is anti aromatic
- 53. Refer theory
- 54. $4 \text{ AI} + 6 \text{ ZnO} \longrightarrow 2 \text{ AI}_2 \text{O}_3 + 6 \text{ Zn}$ ΔH for the above reaction is -ve.
- 55. Due to weak metallic bonding.
- 56. Correct order of basic strength is $NH_2(Et)_2 > EtNH_2 > NMC_3 > Ph NH CH_3$
- 57. 2nd electron gain enthalpy of oxygen is positive.

58.
$$d = \frac{ZM}{N_a a^3}$$
$$= \frac{4 \times 63.55}{6.023 \times 10^{23} \times (x \times 10^{-8})^3} = \frac{422}{x^3} \text{gm/cm}^3$$

59. Let moles of H₂O separated as ice = x gm $\Delta T_f = iK_f m$ $10 = 1 \times 1.86 \ \frac{\frac{62}{62}}{250-x}$

$$x = 64 \text{ gm}$$

- $\begin{array}{cccc} \text{60.} & \text{L}_1 & \text{L}_2 & \text{L}_3 \\ & \text{Green} & \text{Blue} & \text{Red absorbed wave length} \\ & \text{Order of } \lambda \text{ Red > Green > Blue} \\ & \text{L}_3 > \text{L}_1 > \text{L}_2 \end{array}$
 - \therefore Strength of ligand α Δ α 1/ λ
 - \therefore Strength of ligand $L_2 > L_1 > L_3$

PART C - MATHEMATICS

$$\begin{aligned} 61. \qquad & T_n = \frac{\left(3 + \left(n - 1\right) \times 3\right) \left(1^2 + 2^2 + \dots + n^2\right)}{\left(2n + 1\right)} \\ & T_n = \frac{3. \frac{n^2 \left(n + 1\right) \left(2n + 1\right)}{6}}{2n + 1} = \frac{n^2 \left(n + 1\right)}{2} \\ & S_{15} = \frac{1}{2} \sum_{n=1}^{15} \left(n^3 + n^2\right) = \frac{1}{2} \left[\left(\frac{15 \left(15 + 1\right)}{2}\right)^2 + \frac{15 \times 16 \times 31}{6} \right] \\ & = 7820 \end{aligned}$$

62.
$$\lim_{x \to 0^{+}} \frac{x([x] + |x|)\sin[x]}{|x|}$$

$$x \to 0^{-}$$

$$[x] = -1$$

$$|x| = -x$$

$$\Rightarrow \lim_{x \to 0^{-}} \frac{x(-x - 1)\sin(-1)}{-x} = -\sin 1$$

63.
$$f(xy) = f(x).f(y)$$

$$f(0) = 1 \text{ as } f(0) \neq 0$$

$$\Rightarrow f(x) = 1$$

$$\frac{dy}{dx} = f(x) = 1$$

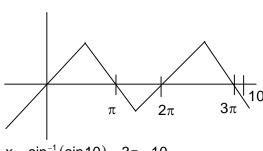
$$\Rightarrow y = x + c$$

$$At, x = 0, y = 1 \Rightarrow c = 1$$

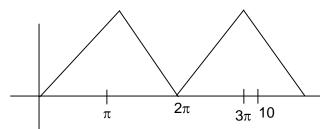
$$y = x + 1$$

$$\Rightarrow y\left(\frac{1}{4}\right) + y\left(\frac{3}{4}\right) = \frac{1}{4} + 1 + \frac{3}{4} + 1 = 3$$

64.



$$x = \sin^{-1}(\sin 10) = 3\pi - 10$$



$$y = \cos^{-1}(\cos 10) = 4\pi - 10$$

 $y - x = \pi$

65.
$$\sin x - \sin 2x + \sin 3x = 0$$

$$\Rightarrow \left(\sin x + \sin 3x\right) - \sin 2x = 0$$

$$\Rightarrow$$
 2 sin x.cos x - sin 2x = 0

$$\Rightarrow$$
 sin 2x (2 cos x - 1) = 0

$$\Rightarrow \sin 2x = 0 \text{ or } \cos x = \frac{1}{2} \Rightarrow x = 0, \frac{\pi}{3}$$

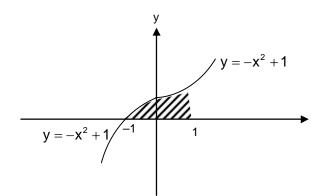
66.
$$z_0 = \omega$$
 or ω^2 (where ω is a non – real cube root of unity)

$$z=3+6i\big(\omega\big)^{81}-3i\big(\omega\big)^{93}$$

$$z = 3 + 3i$$

$$\Rightarrow$$
 arg z = $\frac{\pi}{4}$

$$\int_{-1}^{0} \left(-x^{2} + 1 \right) dx + \int_{0}^{1} \left(x^{2} + 1 \right) dx = 2$$



68.
$$P_1 = x - 4y + 7z - g = 0$$

$$P_2 = 3x - 5y - h = 0$$

$$P_3 = -2x + 5y - 9z - k = 0$$

Here
$$\Delta = 0$$

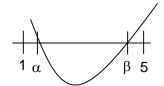
$$2P_1 + P_2 + P_3 = 0$$
 when $2g + h + k = 0$

69.
$$(1-t^6)^3(1-t)^{-3}$$

$$(1-t^{18}-3t^6+3t^{12})(1-t)^{-3}$$

$$\Rightarrow$$
 coefficient of $\,t^4\,$ in $\left(1-t\right)^{\!-3}\,$ is $^{^{3+4-1}}C_4^{}={}^6C_2^{}=15$

70.
$$x^2 - mx + 4 = 0$$
$$\alpha, \beta \in \begin{bmatrix} 1,5 \end{bmatrix}$$



(1)
$$D > 0 \Rightarrow m^2 - 16 > 0$$
$$\Rightarrow m \in (-\infty, -4) \cup (4, \infty)$$

(2)
$$f(1) \ge 0 \Rightarrow 5 - m \ge 0 \Rightarrow m \in (-\infty, 5]$$

$$(3) \qquad f\left(5\right) \geq 0 \Rightarrow 29 - 5m \geq 0 \Rightarrow m \in \left(-\infty, \frac{29}{5}\right]$$

$$(4) 1 < \frac{-b}{2a} < 5 \Rightarrow 1 < \frac{m}{2} < 5 \Rightarrow m \in (2,10)$$
$$\Rightarrow m \in (4,5)$$

No option correct: Bonus

71. Let
$$A(\alpha,0)$$
 and $B(0,\beta)$ be the vectors of the given triangle AOB

$$\Rightarrow |\alpha\beta| = 100$$

$$= 4 \times 9 = 36$$

72.
$$a = A + 6d$$

$$b = A + 10d$$

$$c = A + 12d$$

$$\Rightarrow (A+10d)^2 = (A+6d)(a+12d)$$

$$\Rightarrow \frac{A}{d} = -14$$

$$\frac{a}{c} = \frac{A + 6d}{A + 12d} = \frac{6 + \frac{A}{d}}{12 + \frac{A}{d}} = \frac{6 - 14}{12 - 14} = 4$$

73.
$$\left[\sim (\sim p \lor q) \land (p \land r) \right] \cap (\sim q \land r)$$

$$\equiv \lceil (p \land \sim q) \lor (p \land r) \rceil \land (\sim q \land r)$$

$$\equiv \left[p \wedge (\sim q \vee r) \right] \wedge (\sim q \wedge r)$$

$$\equiv p \wedge (\sim q \wedge r)$$

$$\equiv (p \wedge r) \sim q$$

^{*} If we consider $\alpha, \beta \in (1, 5)$ then option (1) is correct.

74. Vector along the normal to the plane containing the lines $\frac{x}{3} = \frac{y}{4} = \frac{z}{2}$ and $\frac{x}{4} = \frac{y}{2} = \frac{z}{3}$ is $(8\hat{i} - \hat{j} - 10\hat{k})$.

Vector perpendicular to the vectors $2\hat{i} + 3\hat{j} + 4\hat{k}$ and $8\hat{i} - \hat{j} - 10\hat{k}$ is $26\hat{i} - 52\hat{j} + 26\hat{k}$ So, required plane is $26x - 52y + 26z = 0 \implies x - 2y + z = 0$

$$\Rightarrow \frac{\sum x_i^2}{n} = 6$$

(1) . (2)
$$\Rightarrow 4\sum x_i = 4n$$

$$\Longrightarrow \sum x_{_{i}}=n$$

$$\Rightarrow \frac{\sum x_i}{n} = 1$$

$$\Rightarrow$$
 variance = 6 - 1 = 5

$$\Rightarrow$$
 standard diviation = $\sqrt{5}$

- 76. $|A| = e^{-t} \begin{vmatrix} 1 & \cos t & \sin t \\ 1 & -\cos t \sin t & -\sin t + \cos t \\ 1 & 2\sin t & -2\cos t \end{vmatrix}$ $= e^{-t} \left[5\cos^2 t + 5\sin^2 t \right] \forall t \in R$ $= 5e^{-t} \neq 0 \ \forall t \in R$
- 77. $\int \frac{5x^8 + 7x^6}{\left(x^2 + 1 + 2x^7\right)^2} dx$ $= \int \frac{5x^{-6} + 7x^{-8}}{\left(\frac{1}{x^7} + \frac{1}{x^5} + 2\right)^2} dx = \frac{1}{2 + \frac{1}{x^5} + \frac{1}{x^7}} + C$ As f(0) = 0, $f(x) = \frac{x^7}{2x^7 + x^2 + 1}$ $f(1) = \frac{1}{4}$
- 78. $\left| f(x) f(y) \right| \le 2 \left| x y \right|^{3/2}$ divide both side by $\left| x y \right|$ $\left| \frac{f(x) f(y)}{x y} \right| \le 2 \cdot \left| x y \right|^{1/2}$

Apply limit
$$x \to y$$

 $|f'(y)| \le 0 \Rightarrow f'(y) = 0 \Rightarrow f(y) = c \Rightarrow f(x) = 1$

$$\int_{0}^{1} 1. dx = 1$$

79.
$$\frac{dx}{dt} = 3\sec^2 t$$

$$\frac{dy}{dt} = 3 \sec t \tan t$$

$$\frac{dy}{dx} = \frac{\tan t}{\sec t} = \sin t$$

$$\frac{d^2y}{dx^2} = \cos t \frac{dt}{dx}$$

$$= \frac{\cos t}{3 \sec^2 t} = \frac{\cos^3 t}{3} = \frac{1}{3.2\sqrt{2}} = \frac{1}{6\sqrt{2}}$$

80.
$$a_1 a_2 a_3$$

Number of numbers $= 5^3 - 1$

$$\begin{vmatrix} \mathbf{a}_4 & \mathbf{a}_1 & \mathbf{a}_2 & \mathbf{a}_3 \end{vmatrix}$$

2 ways for a₄

Numbers of numbers = 2×5^3

Required number 0020 = $5^3 + 2 \times 5^3 - 1$

81.
$$x^2 + y^2 - 16x - 20y + 164 = r^2$$

$$A(8,10), R_1 = r$$

$$(x-4)^2 + (y-7)^2 = 36$$

B (4, 7),
$$R_2 = 6$$

$$|R_1 - R_2| < AB < R_1 + R_2$$

$$\Rightarrow$$
 1 < r < 11

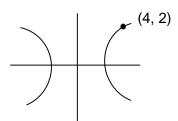
82. Given hyperbola is

$$\frac{x^2}{4} - \frac{y^2}{b^2} = 1$$

Satisfying the point (4, 2)

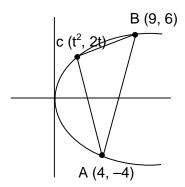
$$\Longrightarrow b^2 = \frac{4}{3}$$

$$\Rightarrow$$
 e = $\frac{2}{\sqrt{3}}$



83. For maximum area, tangent at the point c must be parallel to chord BC.

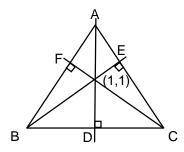
$$\therefore t = \frac{1}{2}$$



84. Equation of AB is 3x - 2y + 6 = 0Equation of AC is 4x + 5y - 20 = 0. Equation of BE is 2x + 3y - 5 = 0

Equation of CF is
$$5x - 4y - 1 = 0$$

$$26x - 122y = 1675$$



85. E₁: Event of drawing a Red ball and placing a green ball in the bag

E2: Event of drawing a green ball and placing a red ball in the bag

E : Event of drawing a red ball in second draw $P(E) = P(E_1) \times P\left(\frac{E}{E_1}\right) + P(E_2) \times P\left(\frac{E}{E_2}\right)$

$$=\frac{5}{7}\times\frac{4}{7}+\frac{2}{7}\times\frac{6}{7}=\frac{32}{49}$$

86. Line x = ay + b, z = cy + d

$$\Rightarrow \frac{x-b}{a} = \frac{y}{1} = \frac{z-d}{c}$$

Line x = a'z + b', y = c'z + d'

$$\Rightarrow \frac{x-b'}{a'} = \frac{y-d'}{c'} = \frac{z}{1}$$

Given both the lines are perpendicular

$$\Rightarrow$$
 aa'+ c'+ c = 0

87. Projection of \vec{b} on $\vec{a} = \frac{\vec{a} \cdot \vec{b}}{|\vec{a}|} = |\vec{a}|$

$$\Rightarrow b_1 + b_2 = 2 \qquad \dots (1)$$

and
$$(\vec{a} + \vec{b}) \perp \vec{c} \Rightarrow (\vec{a} + \vec{b}) \cdot \vec{c} = 0$$

$$\Rightarrow 5b_1 + b_2 = -10 \qquad \dots (2)$$

from (1) and (2) \Rightarrow b₁ = -3 and b₂ = 5

then
$$|\vec{b}| = \sqrt{b_1^2 + b_2^2 + 2} = 6$$

88. D must be perfect square

$$\Rightarrow$$
 121 – 24 $\alpha = \lambda^2$

 \Rightarrow maximum value of α is 5

$$\alpha = 1 \Longrightarrow \lambda \notin I$$

$$\alpha=2\Longrightarrow \lambda\not\in I$$

$$\alpha=3\Rightarrow\lambda\in I$$

⇒ 3 integral values

$$\alpha=4 \Longrightarrow \lambda \in I$$

$$\alpha = 5 \Rightarrow \lambda \in I$$

89. $f(x) = 2\left(1 + \frac{1}{x-1}\right)$

$$f'(x) = -\frac{2}{(x-1)^2}$$

 \Rightarrow f is one – one but not onto

 $90. \qquad \frac{1}{\sqrt{2k}} \int\limits_0^{\pi/3} \frac{\tan \theta}{\sqrt{\sec \theta}} \, d\theta = \frac{1}{\sqrt{2k}} \int\limits_0^{\pi/3} \frac{\sin \theta}{\sqrt{\cos \theta}} \, d\theta$

$$= -\frac{1}{\sqrt{2k}} 2\sqrt{\cos\theta} \Big|_0^{\pi/3} = -\frac{\sqrt{2}}{\sqrt{k}} \left(\frac{1}{\sqrt{2}} - 1\right)$$

$$\Rightarrow$$
 k = 2